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Remarks:

This application was filed on 22 - 12 - 1999 as a  
divisional application to the application mentioned  
under INID code 62.

(54) **Liquid crystal all purpose cleaning compositions**

(57) An improvement is described in a liquid crystal  
composition which is more environmentally friendly. The  
liquid crystal cleaning composition is especially effective  
in the removal of oily and greasy soil and has an evidenced  
grease release effect, contains an anionic detergent, an  
ethoxylated glycerol type compound, a hydrocarbon ingredient,  
and water.

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**Description**

[0001] The present application is a divisional application of European patent application 95 939 106.1, the contents thereof being incorporated herein by reference.

**Field of the invention**

[0002] This invention relates to an improved all-purpose liquid cleaner in the form of a liquid crystal or a microemulsion designed in particular for cleaning hard surfaces and which is effective in removing grease soil and/or bath soil and in leaving unrinsed surfaces with a shiny appearance.

**Background of the invention**

[0003] In recent years all-purpose liquid detergents have become widely accepted for cleaning hard surfaces, e.g. painted woodwork and panels, tiled walls, wash bowls, bathtubs, linoleum or tile floors, washable wall paper, etc.. Such all-purpose liquids comprise clear and opaque aqueous mixtures of water-soluble synthetic organic detergents and water-soluble detergent builder salts. In order to achieve comparable cleaning efficiency with granular or powdered all-purpose cleaning compositions, use of water-soluble inorganic phosphate builder salts was favored in the prior art all-purpose liquids. For example, such early phosphate-containing compositions are described in U.S. Patent Nos

[0004] In view of the environmentalist's efforts to reduce phosphate levels in ground water, improved all-purpose liquids containing reduced concentrations of inorganic phosphate builder salts or non-phosphate builder salts have appeared. A particularly useful self-opacified liquid of the latter type is described in U.S. Patent No. 4,244,840.

[0005] However, these prior art all-purpose liquid detergents containing detergent builder salts or other equivalent tend to leave films, spots or streaks on cleaned unrinsed surfaces, particularly shiny surfaces. Thus, such liquids require thorough rinsing of the cleaned surfaces which is a time-consuming chore for the user.

[0006] In order to overcome the foregoing disadvantage of the prior art all-purpose liquid, U.S. Patent No. 4,017,409 teaches that a mixture of paraffin sulfonate and a reduced concentration of inorganic phosphate builder salt should be employed. However, such compositions are not completely acceptable from an environmental point of view based upon the phosphate content. On the other hand, another alternative to achieving phosphate-free all-purpose liquids has been to use a major proportion of a mixture of anionic and nonionic detergents with minor amounts of glycol ether solvent and organic amine as shown in U.S. Patent NO. 3,935,130. Again, this approach has not been completely satisfactory and the high levels of organic detergents necessary to achieve cleaning cause foaming which, in turn, leads to the need for thorough rinsing which has been found to be undesirable to today's consumers.

[0007] Another approach to formulating hard surfaced or all-purpose liquid detergent composition where product homogeneity and clarity are important considerations involves the formation of oil-in-water (o/w) microemulsions which contain one or more surface-active detergent compounds, a water-immiscible solvent (typically a hydrocarbon solvent), water and a "cosurfactant" compound which provides product stability. By definition, an o/w microemulsion is a spontaneously forming colloidal dispersion of "oil" phase particles having a particle size in the range of 25 Å to 800 Å in a continuous aqueous phase.

[0008] In view of the extremely fine particle size of the dispersed oil phase particles, microemulsions are transparent to light and are clear and usually highly stable against phase separation.

[0009] Patent disclosures relating to use of grease-removal solvents in o/w microemulsions include, for example, European Patent Applications EP 0137615 and EP 0137616 - Herbots et al; European Patent Application EP 0160762 - Johnston et al; and U.S. Patent No. 4,561,991 - Herbots et al. Each of these patent disclosures also teaches using at least 5% by weight of grease-removal solvent.

[0010] It also is known from British Patent Application GB 2144763A to Herbots et al, published March 13, 1985, that magnesium salts enhance grease-removal performance of organic grease-removal solvents, such as the terpenes, in o/w microemulsion liquid detergent compositions. The compositions of this invention described by Herbots et al. require at least 5% of the mixture of grease-removal solvent and magnesium salt and preferably at least 5% of solvent (which may be a mixture of water-immiscible non-polar solvent with a sparingly soluble slightly polar solvent) and at least 0.1% magnesium salt.

[0011] However, since the amount of water immiscible and sparingly soluble components which can be present in an o/w microemulsion, with low total active ingredients without impairing the stability of the microemulsion is rather limited (for example, up to 18% by weight of the aqueous phase), the presence of such high quantities of grease-removal solvent tend to reduce the total amount of greasy or oily soils which can be taken up by and into the microemulsion without causing phase separation.

[0012] The following representative prior art patents also relate to liquid detergent cleaning compositions in the

form of o/w microemulsions: U.S. Patents Nos. 4,472,291 - Rosario; 4,540,448 - Gauter et al; 3,723,330 - Sheflin; etc.

[0013] liquid detergent compositions which include terpenes, such as d-limonene, or other grease-removal solvent, although not disclosed to be in the form of o/w microemulsions, are the subject matter of the following representative patent documents: European Patent Application 0080749; British Patent Specification 1,603,047; 4,414,128; and 4,540,505. For example, U.S. Patent No. 4,414,128 broadly discloses an aqueous liquid detergent composition characterized by, by weight:

- (a) from 1% to 20% of a synthetic anionic, nonionic, amphoteric or zwitterionic surfactant or mixture thereof;
- (b) from 0.5% to 10% of a mono- or sesquiterpene or mixture thereof, at a weight ratio of (a):(b) lying in the range of 5:1 to 1:3; and
- (c) from 0.5% to 10% of a polar solvent having a solubility in water at 15°C in the range of from 0.2% to 10%. Other ingredients present in the formulations disclosed in this patent include from 0.05% to 2% by weight of an alkali metal, ammonium or alkanolammonium soap of a C<sub>13</sub>-C<sub>24</sub> fatty acid; a calcium sequestrant from 0.5% to 13% by weight; non-aqueous solvent, e.g., alcohols and glycol ethers, up to 10% by weight; and hydrotropes, e.g., urea, ethanalamines, salts of lower alkylaryl sulfonates, up to 10% by weight. All of the formulations shown in the Examples of this patent include relatively large amounts of detergent builder salts which are detrimental to surface shine.

[0014] Furthermore, the present inventors have observed that in formulations containing grease-removal assisting magnesium compounds, the addition of minor amounts of builder salts, such as alkali metal polyphosphates, alkali metal carbonates, nitrilotriacetic acid salts, and so on, tends to make it more difficult to form stable microemulsion systems.

[0015] U.S. Patent 5,082,584 discloses a microemulsion composition having an anionic surfactant, a cosurfactant, nonionic surfactant, perfume and water; however, these compositions do not possess the low ecotoxicity profile and the improved interfacial tension properties as exhibited by the compositions of the instant invention.

[0016] British Patent No 1,453,385 discloses polyesterified nonionic surfactants similar to the polyesterified nonionic surfactants of the instant invention. However, these nonionic surfactants of British Patent 1,453,385 do not disclose the formula (II) portion of the instant composition. Additionally, the formulated compositions of British Patent 1,453,385 fail to disclose the critical limitations of the instant invention.

[0017] A number of patents teach esterified ethoxylated glycerol compounds for various applications. These patents are Great Britain 1,453,385; Japan 59-1600 and Japan 58-206693 and European Patent Application 0586,323A1. These publications fail to appreciate that a mixture of esterified ethoxylated glycerol and nonesterified ethoxylated glycerol, when used in a hard surface cleaning composition, functions as a grease release agent.

#### Summary of the Invention

[0018] The present invention provides an improved, clear, liquid cleaning composition having improved interfacial tension which improves cleaning hard surface in the form of a liquid crystal or a microemulsion which is suitable for cleaning hard surfaces such as plastic, vitreous and metal surfaces having a shiny finish. More particularly, the improved cleaning compositions exhibit good grease soil removal properties due to the improved interfacial tensions, when used in undiluted (neat) form and leave the cleaned surfaces shiny without the need of or requiring only minimal additional rinsing or wiping. The latter characteristic is evidenced by little or no visible residues on the unrinsed cleaned surfaces and, accordingly, overcomes one of the disadvantages of prior art products. The instant compositions exhibit a grease release effect in that the instant compositions impede or decrease the anchoring of greasy soil on surfaces that have been cleaned with the instant compositions as compared to surfaces cleaned with a liquid crystal composition or a commercial microemulsion composition which means that the grease soiled surface is easier to clean upon subsequent cleanings. Surprisingly, these desirable results are accomplished even in the absence of polyphosphate or other inorganic or organic detergent builder salts and also in the complete absence or substantially complete absence of grease-removal solvent.

[0019] The instant compositions are more friendly for the environment due to the low ecotoxicity of the ethoxylated glycerol type compounds used in the instant compositions.

[0020] The compositions of the instant invention have an ecotoxicity value as measured by the LC 50 test as deferred by The Organization for Economic Cooperation and Development (OECD) (of which the United States is a member) in OECD Test No. 202 of at least 0.18 ml/L measured on Daphniae microorganisms.

[0021] In one aspect, the invention generally provides a stable, clear all-purpose, hard surface cleaning composition especially effective in the removal of oily and greasy oil, which is in the form of a substantially dilute oil-in-water microemulsion having an aqueous phase and an oil phase. The dilute o/w microemulsion comprises on a weight basis:

from 0.1% to 20% of an anionic surfactant;

from 0.1% to 50% of a water-mixable cosurfactant having either limited ability or substantially no ability to dissolve oily or greasy soil;

0.1% to 20% of a compound which is a mixture of a partially esterified ethoxylated polyhydric alcohol, a fully esterified ethoxylated polyhydric alcohol and a nonesterified ethoxylated polyhydric alcohol, said mixture being (herein after referred to as an ethoxylated glycerol type compound);

0 to 15% of magnesium sulfate heptahydrate;

0.1 to 10.0% of a perfume or water insoluble hydrocarbon; and

10 to 85% of water, said proportions being based upon the total weight of the composition.

[0022] In another aspect, the invention generally provides a stable, clear all-purpose, hard surface cleaning composition especially effective in the removal of oily and greasy oil, which is in the form of a substantially dilute oil-in-water microemulsion having an aqueous phase and an oil phase. The aqueous phase of the dilute o/w microemulsion comprises on a weight basis:

from 0.1 % to 20% of an anionic surfactant;

from 0.1% to 50% of a water-mixable cosurfactant having either limited ability or substantially no ability to dissolve oily or greasy soil;

0.4% to 1.0% of a trialkyl ester of citric acid;

0.1% to 10% of a mixture of a partially esterified ethoxylated polyhydric alcohol, a fully esterified ethoxylated polyhydric alcohol and a nonesterified polyhydric alcohol, (said mixture being herein after referred to as an ethoxylated glycerol type compound);

0 to 15% of magnesium sulfate heptahydrate;

0.4 to 10.0% of a perfume or water insoluble hydrocarbon; and

10 to 85% of water, said proportions being based upon the total weight of the composition.

[0023] In another aspect, the invention generally provides a stable, clear all-purpose, hard surface cleaning composition especially effective in the removal of particle soil, which is in the form of a substantially dilute oil-in-water microemulsion having an aqueous phase and an oil phase. The dilute o/w microemulsion comprises on a weight basis:

from 0.1% to 20% of an anionic surfactant;

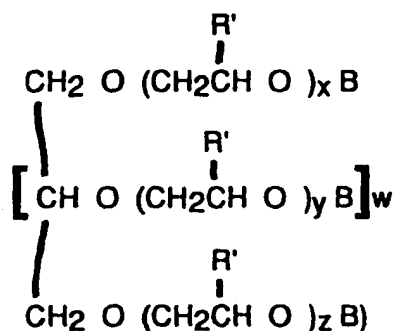
from 0.1% to 50% of a water-mixable cosurfactant having either limited ability or substantially no ability to dissolve oily or greasy soil;

0.1% to 20% of an ethoxylated polyhydric alcohol;

0 to 15% of magnesium sulfate heptahydrate;

0.1 to 10.0% of a perfume or water insoluble hydrocarbon; and

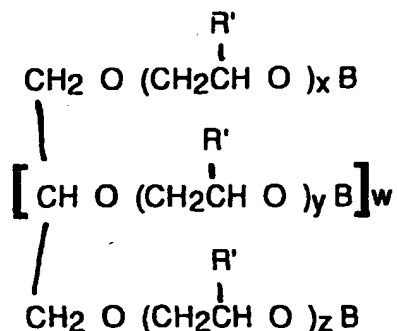
10 to 85% of water, said proportions being based upon the total weight of the composition. This composition can also contain 0 to 10 wt. % of a monoester of an ethoxylated polyhydric alcohol depicted by the formula:



wherein w equals one to four. Two of the B's are hydrogen and one B is selected from the group consisting of a group represented by:



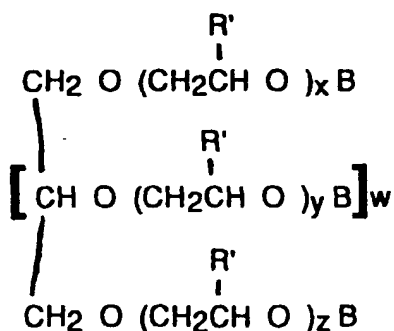
wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, more preferably 11 to 15 carbon atoms and alkenyl groups having 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals about 2 to about 100. This composition can also contain 0 to 2 wt. %, of a diester of an ethoxylated polyhydric alcohol depicted by the formula



wherein w equals one to four. One of the B's is hydrogen and two B's are selected from the group consisting of a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100. This composition can also contain 0 to 1.0 wt. %, of a triester of an ethoxylated polyhydric alcohol depicted by the formula



wherein w equals one to four. The three B's are selected from the group consisting of a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals about 2 to about 100.

[0024] The dispersed oil phase of the o/w microemulsion is composed essentially of the water-immiscible or hardly water-soluble perfume.

[0025] Quite surprisingly although the perfume is not, per se, a solvent for greasy or oily soil, -- even though some perfumes may, in fact, contain as much as 80% of terpenes which are known as good grease solvents -- the inventive compositions in dilute form have the capacity to solubilize up to 10 times or more of the weight of the perfume of oily and greasy soil, which is removed or loosened from the hard surface by virtue of the action of the anionic and nonionic surfactants, said soil being taken up into the oil phase of the o/w microemulsion.

[0026] In another aspect, the invention generally provides highly concentration microemulsion compositions in the form of either an oil-in-water (o/w) microemulsion or a water-in-oil (w/o) microemulsion which when diluted with additional water before use can form dilute o/w microemulsion compositions. The concentrated microemulsion compositions comprise, by weight, 0.1% to 20% of an anionic surfactant, 0.1% to 20% of an ethoxylated glycerol type compound, 0% to 2.5% of a fatty acid, 0.1% to 10% of perfume or water insoluble hydrocarbon having 6 to 18 carbon atoms, 0.1% to 50% of a cosurfactant, and 20% to 97% of water.

[0027] In another aspect of the invention, liquid crystal compositions are provided which comprise by weight 0.1% to 20% of an anionic surfactant, 0.1% to 20% of an ethoxylated glycerol type compound, 0 to 2.5% of a fatty acid, 0.1% to 10% of a perfume, more preferably 1% to 10%, 1% to 50% of cosurfactant selected from the group consisting of propylene glycol monobutylether, dipropylene glycol monobutylether and tripropyleneglycol monobutyl ether and mixtures thereof and the balance being water.

[0028] In another aspect, the invention provides highly concentration microemulsion compositions in the form of either an oil-in-water (o/w) microemulsion or a water-in-oil (w/o) microemulsion which when diluted with additional water before use can form dilute o/w microemulsion compositions. The concentrated microemulsion compositions comprise, by weight, 0.1% to 20% of an anionic surfactant, 0.1% to 20% of an ethoxylated glycerol type compound, 0% to 2.5% of a fatty acid, 0.1% to 10% of perfume or water insoluble hydrocarbon having 6 to 18 carbon atoms, 0.1% to 50% of a cosurfactant, and 20% to 97% of water.

[0029] In another aspect, the invention generally provides highly concentration microemulsion compositions in the form of either an oil-in-water (o/w) microemulsion or a water-in-oil (w/o) microemulsion which when diluted with additional water before use can form dilute o/w microemulsion compositions. The concentrated microemulsion compositions comprise, by weight, 0.1% to 20% of an anionic surfactant, 0.1% to 20% of an ethoxylated polyhydric alcohol, 0.1% to 10% of perfume or water insoluble hydrocarbon having 6 to 18 carbon atoms, 0.1% to 50% of a cosurfactant, and 20% to 97% of water.

[0030] In another aspect of the invention, liquid crystal compositions are provided which comprise by weight 0.1% to 20% of an anionic surfactant, 0.1% to 20% of an ethoxylated polyhydric alcohol, 0.1% to 10% of a perfume, 1% to 50% of cosurfactant and the balance being water.

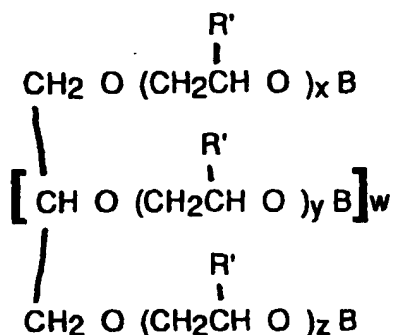
#### Detailed Description of the Invention

[0031] The present invention relates to a stable liquid crystal or microemulsion composition comprising by weight: 0.1% to 20% of an anionic surfactant, 0.1% to 50% of a cosurfactant, 0.1% to 20% of an ethoxylated glycerol type compound, 0.1% to 10% of a water insoluble hydrocarbon or a perfume and the balance being water, said composition having an ecotoxicity value as measured by the LC50 test of at least 0.18 ml/L measured on Daphniae microorganisms.

[0032] The present invention relates to a stable liquid crystal or microemulsion composition comprising by weight: 0.1% to 20% of an anionic surfactant, 0.1% to 50% of a cosurfactant, 0.4 to 1.0 wt. % of a trialkyl ester of citric acid such as tri-n butyl citrate, 0.1% to 20% of an ethoxylated glycerol type compound, 0.1% to 10% of a water insoluble hydrocarbon or a perfume and the balance being water, said composition having an ecotoxicity value as measured by the LC50 test of at least 0.18 ml/L measured on Daphniae microorganisms.

[0033] The present invention relates to a stable liquid crystal or microemulsion composition comprising by weight: 0.1% to 20% of an anionic surfactant, 0.1% to 50% of a cosurfactant, 0.1% to 20% of an ethoxylated polyhydric alcohol, 0.1% to 10% of a water insoluble hydrocarbon or a perfume and the balance being water. This composition can also

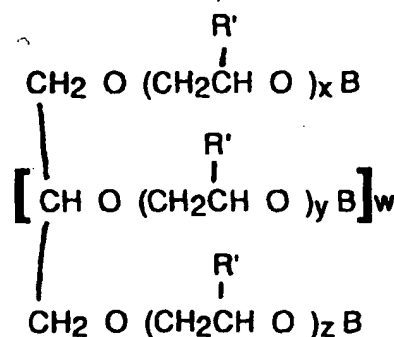
contain 0 to 10 wt. % of a monoester of an ethoxylated polyhydric alcohol depicted by the formula



wherein w equals one to four. Two of the B's are hydrogen and one B is selected from the group consisting of a group represented by:



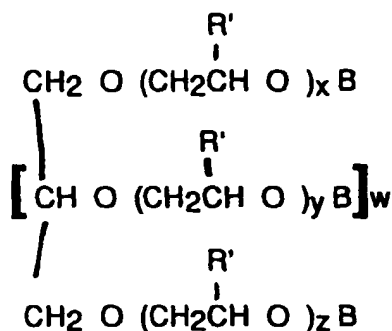
wherein R is selected from the group consisting of alkyl group having about 6 to 22 carbon atoms, and alkenyl groups having about 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals about 2 to about 100. This composition can also contain 0 to 2 wt. % of a diester of an ethoxylated polyhydric alcohol depicted by the formula



wherein w equals one to four. One of the B's is hydrogen and two B's are selected from the group consisting of a group represented by:



wherein R is selected from the group consisting of alkyl group having about 6 to 22 carbon atoms, and alkenyl groups having about 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals about 2 to about 100. This composition can also contain 0 to about 1.0 wt. % of a triester of an ethoxylated polyhydric alcohol depicted by the formula



wherein w equals one to four. The three B's are selected from the group consisting of a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100.

[0034] According to the present invention, the role of the hydrocarbon is provided by a non-water-soluble perfume. Typically, in aqueous based compositions the presence of a solubilizers, such as alkali metal lower alkyl aryl sulfonate hydrotrope, triethanolamine, urea, etc., is required for perfume dissolution, especially at perfume levels of 1% and higher, since perfumes are generally a mixture of fragrant essential oils and aromatic compounds which are generally not water-soluble. Therefore, by incorporating the perfume into the aqueous cleaning composition as the oil (hydrocarbon) phase of the ultimate o/w microemulsion composition, several different important advantages are achieved.

[0035] First, the cosmetic properties of the ultimate cleaning composition are improved: the compositions are both clear (as a consequence of the formation of a microemulsion) and highly fragranced (as a consequence of the perfume level).

[0036] Second, the need for use of solubilizers, which do not contribute to cleaning performance, is eliminated.

[0037] Third, an improved grease release effect and an improved grease removal capacity in neat (undiluted) usage of the dilute aspect or after dilution of the concentrate can be obtained without detergent builders or buffers or conventional grease removal solvents at neutral or acidic pH and at low levels of active ingredients while improved cleaning performance can also be achieved in diluted usage.

[0038] As used herein and in the appended claims the term "perfume" is used in its ordinary sense to refer to and include any non-water soluble fragrant substance or mixture of substances including natural (i.e., obtained by extraction of flower, herb, blossom or plant), artificial (i.e., mixture of natural oils or oil constituents) and synthetically produced substance) odoriferous substances. Typically, perfumes are complex mixtures of blends of various organic compounds such as alcohols, aldehydes, ethers, aromatic compounds and varying amounts of essential oils (e.g., terpenes) such as from 0% to 80%, by weight, the essential oils themselves being volatile odoriferous compounds and also serving to dissolve the other components of the perfume.

[0039] In the present invention the precise composition of the perfume is of no particular consequence to cleaning performance so long as it meets the criteria of water immiscibility and having a pleasing odor. Naturally, of course, especially for cleaning compositions intended for use in the home, the perfume, as well as all other ingredients, should be cosmetically acceptable, i.e., non-toxic, hypoallergenic, etc.. The instant compositions show a marked improvement in ecotoxicity as compared to existing commercial products.

[0040] The hydrocarbon such as a perfume is present in the dilute o/w microemulsion in an amount of from 0.1% to 10% by weight. If the amount of hydrocarbon (perfume) is less than 0.4% by weight it becomes more difficult to form the o/w microemulsion. In the case of the liquid crystal one need at least 0.5 weight % of perfume. If the hydrocarbon (perfume) is added in amounts more than 10% by weight, the cost is increased without any additional cleaning benefit and, in fact, with some diminishing of cleaning performance insofar as the total amount of greasy or oily soil which can be taken up in the oil phase of the microemulsion will decrease proportionately.

[0041] Furthermore, although superior grease removal performance will be achieved for perfume compositions not

containing any terpene solvents, it is apparently difficult for perfumers to formulate sufficiently inexpensive perfume compositions for products of this type (i.e., very cost sensitive consumer-type products) which includes less than 20%, usually less than 30%, of such terpene solvents.

[0042] Thus, merely as a practical matter, based on economic consideration, the dilute o/w microemulsion detergent cleaning compositions of the present invention may often include as much as 0.2% to 7% by weight, based on the total composition, of terpene solvents introduced thereunto via the perfume component. However, even when the amount of terpene solvent in the cleaning formulation is less than 1.5% by weight, such as up to 0.6% by weight or 0.4% by weight or less, satisfactory grease removal and oil removal capacity is provided by the inventive diluted o/w microemulsions.

[0043] Thus, for a typical formulation of a diluted o/w microemulsion according to this invention a 20 milliliter sample of o/w microemulsion containing 1% by weight of perfume will be able to solubilize, for example, up to 2 to 3 ml of greasy and/or oily soil, while retaining its form as a microemulsion, regardless of whether the perfume contains 0%, 0.1%, 0.2%, 0.3%, 0.4%, 0.5%, 0.6%, 0.7% or 0.8% by weight of terpene solvent. In other words, it is an essential feature of the compositions of this invention that grease removal is a function of the result of the microemulsion, per se, and not of the presence or absence in the microemulsion of a "greasy soil removal" type of solvent.

[0044] In place of the perfume one can employ a water insoluble paraffin or isoparaffin having 6 to 18 carbon at a concentration of 0.4 to 8.0 wt. percent.

[0045] The water-soluble organic detergent materials which are used in forming the ultimate o/w microemulsion compositions of this invention may be selected from the group consisting of water-soluble, non-soap, anionic surfactants mixed with a fatty acid and the solubilizing agent which is a partially esterified ethoxylated polyhydric alcohol such as a partially esterified ethoxylated glycerol.

[0046] Although conventional nonionic surfactants can be used in the instant compositions, the employment of such conventional nonionic in the instant composition will decrease the environmental profile of the composition as well as having an adverse effect on the grease release and grease + particulate soil removal properties of the composition.

[0047] Regarding the anionic surfactant present in the o/w microemulsions any of the conventionally used water-soluble anionic surfactants or mixtures of said anionic detergents and anionic detergents can be used in this invention. As used herein the term "anionic surfactant" is intended to refer to the class of anionic and mixed anionic-nonionic surfactants providing deterative action.

[0048] Suitable water-soluble non-soap, anionic surfactants include those surface-active or detergent compounds which contain an organic hydrophobic group containing generally 8 to 26 carbon atoms in their molecular structure and at least one water-solubilizing group selected from the group of sulfonate, sulfate and carboxylate so as to form a water-soluble surfactant. Usually, the hydrophobic group will include or comprise a C<sub>8</sub>-C<sub>22</sub> alkyl, alkyl or acyl group. Such surfactants are employed in the form of water-soluble salts and the salt-forming cation usually is selected from the group consisting of sodium, potassium, ammonium, magnesium and mono-, di- or tri-C<sub>2</sub>-C<sub>3</sub> alkanolammonium, with the sodium, magnesium and ammonium cations again being preferred.

[0049] Examples of suitable sulfonated anionic surfactants are the well known higher alkyl mononuclear aromatic sulfonates such as the higher alkyl benzene sulfonates containing from 10 to 16 carbon atoms in the higher alkyl group in a straight or branched chain, C<sub>8</sub>-C<sub>15</sub> alkyl toluene sulfonates and C<sub>8</sub>-C<sub>15</sub> alkyl phenol sulfonates.

[0050] A preferred sulfonate is linear alkyl benzene sulfonate having a high content of 3- (or higher) phenyl isomers and a correspondingly low content (well below 50%) of 2- (or lower) phenyl isomers, that is, wherein the benzene ring is preferably attached in large part at the 3 or higher (for example, 4, 5, 6 or 7) position of the alkyl group and the content of the isomers in which the benzene ring is attached in the 2 or 1 position is correspondingly low. Particularly preferred materials are set forth in U.S. Patent 3,320,174.

[0051] Other suitable anionic surfactants are the olefin sulfonates, including long-chain alkene sulfonates, long-chain hydroxyalkane sulfonates or mixtures of alkene sulfonates and hydroxyalkane sulfonates. These olefin sulfonate detergents may be prepared in a known manner by the reaction of sulfur trioxide (SO<sub>3</sub>) with long-chain olefins containing 8 to 25, and having the formula RCH=CHR<sub>1</sub> where R is a higher alkyl group of 6 to 23 carbons and R<sub>1</sub> is an alkyl group of 1 to 17 carbons or hydrogen to form a mixture of sultones and alkene sulfonic acids which is then treated to convert the sultones to sulfonates. Olefin sulfonates contain from 14 to 16 carbon atoms in the R alkyl group and are obtained by sulfonating an α-olefin.

[0052] Other examples of suitable anionic sulfonate surfactants are the paraffin sulfonates containing 10 to 20. Primary paraffin sulfonates are made by reacting long-chain α olefins and bisulfites and paraffin sulfonates having the sulfonate group distributed along the paraffin chain are shown in U.S. Patents Nos. 2,503,280; 2,507,088; 3,260,744; 3,372,188; and German Patent 735,096.

[0053] Examples of satisfactory anionic sulfate surfactants are the C<sub>8</sub>-C<sub>18</sub> alkyl sulfate salts and the C<sub>8</sub>-C<sub>18</sub> alkyl sulfate salts and the C<sub>8</sub>-C<sub>18</sub> alkyl ether polyethenoxy sulfate salts having the formula R(OC<sub>2</sub>H<sub>4</sub>)<sub>n</sub>OSO<sub>3</sub>M wherein n is 1 to 12, and M is a solubilizing cation selected from the group consisting of sodium, potassium, ammonium, magnesium and mono-, di- and triethanol ammonium ions. The alkyl sulfates may be obtained by sulfating the alcohols obtained by

reducing glycerides of coconut oil or tallow or mixtures thereof and neutralizing the resultant product. On the other hand, the alkyl ether polyethenoxy sulfates are obtained by sulfating the condensation product of ethylene oxide with a C<sub>8</sub>-C<sub>18</sub> alkanol and neutralizing the resultant product. The alkyl sulfates may be obtained by sulfating the alcohols obtained by reducing glycerides of coconut oil or tallow or mixtures thereof and neutralizing the resultant product. On the other hand, the alkyl ether polyethenoxy sulfates are obtained by sulfating the condensation product of ethylene oxide with a C<sub>8</sub>-C<sub>18</sub> alkanol and neutralizing the resultant product. The alkyl ether polyethenoxy sulfates differ from one another in the number of moles of ethylene oxide reacted with one mole of alkanol. Preferred alkyl sulfates and preferred alkyl ether polyethenoxy sulfates contain 10 to 16 carbon atoms in the alkyl group.

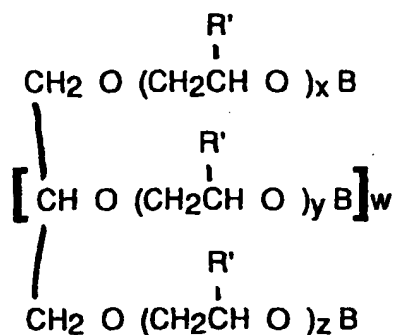
[0054] The C<sub>8</sub>-C<sub>12</sub> alkylphenyl ether polyethenoxy sulfates containing from 2 to 6 moles of ethylene oxide in the molecule also are suitable for use in the inventive compositions. These detergents can be prepared by reacting an alkyl phenol with 2 to 6 moles of ethylene oxide and sulfating and neutralizing the resultant ethoxylated alkylphenol.

[0055] Obviously, these anionic surfactants will be present either in acid form or salt form depending upon the pH of the final composition, with salt forming cation being the same as for the other anionic detergents.

[0056] Of the foregoing non-soap anionic surfactants, the preferred surfactants are the C<sub>9</sub>-C<sub>15</sub> linear alkylbenzene sulfonates and the C<sub>13</sub>-C<sub>17</sub> paraffin or alkane sulfonates. Particularly, preferred compounds are sodium C<sub>10</sub>-C<sub>13</sub> alkylbenzene sulfonate and sodium C<sub>13</sub>-C<sub>17</sub> alkane sulfonate.

[0057] Generally, the proportion of the nonsoap-anionic surfactant will be in the range of 0.1% to 20.0%, by weight of the dilute o/w microemulsion composition.

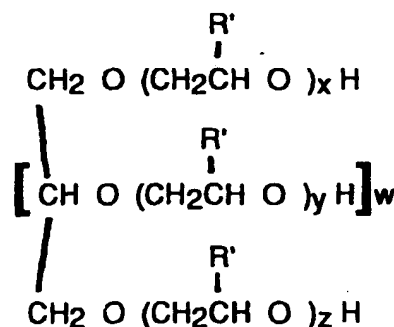
[0058] The instant composition contains a composition (herein after referred to as ethoxylated glycerol type compound) which is a mixture of a fully esterified ethoxylated polyhydric alcohol, a partially esterified ethoxylated polyhydric alcohol and a nonesterified ethoxylated polyhydric alcohol, wherein the preferred polyhydric alcohol is glycerol, and the compound is:



Formula

(I)

and



Formula

(II)

wherein w equals one to four. B is selected from the group consisting of hydrogen or a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, wherein at least one of the B groups is represented by said



and R' is selected from the group consisting of hydrogen and methyl groups; x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100, wherein in Formula (I) the ratio of monoester / diester / triester is 45 to 90 / 5 to 40 / 1 to 20, wherein the ratio of Formula (I) to Formula (II) is a value between 3 to 0.02, wherein it is most preferred that there is more of Formula (II) than Formula (I) in the mixture that forms the compound.

[0059] The ethoxylated glycerol type compound used in the instant composition is manufactured by the Kao Corporation and sold under the trade name LEVENOL™ such as Levenol F-200 which has an average EO of 6 and a molar ratio of coco fatty acid to glycerol of 0.55 or LEVENOL™ V501/2 which has an average EO of 17 and a molar ratio of tallow fatty acid to glycerol of 1.0. It is preferred that the molar ratio of the fatty acid to glycerol is less than 1.7, more preferably less than 1.5 and most preferably less than 1.0. The ethoxylated glycerol type compound has a molecular weight of 400 to 1600, and a pH (50 grams / liter of water) of 5-7. The LEVENOL™ compounds are substantially non irritant to human skin and have a primary biodegradability higher than 90% as measured by the Wickbold method Bias-7d.

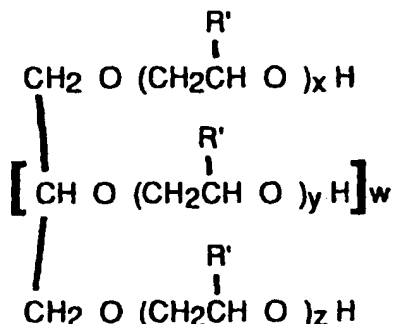
[0060] Two examples of the LEVENOL™ compounds are LEVENOL™ V-501/2 which has 17 ethoxylated groups and is derived from tallow fatty acid with a fatty acid to glycerol ratio of 1.0 and a molecular weight of 1465 and LEVENOL™ F-200 has 6 ethoxylated groups and is derived from coco fatty acid with a fatty acid to glycerol ratio of 0.55. Both LEVENOL™ F-200 and Levenol™ V-501/2 are composed of a mixture of Formula (I) and Formula (II). The LEVENOL™ compounds has ecotoxicity values of algae growth inhibition > 100 mg/liter; acute toxicity for Daphniae > 100 mg/liter and acute fish toxicity > 100 mg/liter. The LEVENOL™ compounds have a ready biodegradability higher than 60% which is the minimum required value according to OECD 301B measurement to be acceptably biodegradable.

[0061] Polyesterified nonionic compounds also useful in the instant compositions are CROVOL™ PK-40 and CROVOL™ PK-70 manufactured by Croda GMBH of the Netherlands. CROVOL™ PK-40 is a polyoxyethylene (12) Palm Kernel Glyceride which has 12 EO groups. CROVOL™ PK-70 which is preferred is a polyoxyethylene (45) Palm Kernel Glyceride have 45 EO groups.

[0062] In the dilute o/w microemulsion compositions or liquid crystal compositions the ethoxylated glycerol type compounds or the polyesterified nonionic compounds will be present in admixture with the anionic detergent. The proportion of the ethoxylated glycerol type compound or the polyesterified nonionic solubilizing agent based upon the weight of the liquid crystal composition or the final dilute o/w microemulsion composition will be 0.1% to 20%, by weight.

[0063] Furthermore, in the more preferred compositions the weight ratio of nonsoap anionic detergent to the ethoxylated glycerol type compound will be in the range of 3:1 to 1:3 with especially good results being obtained at a weight ratio of 2:1.

[0064] The ethoxylated polyhydric alcohol such as an ethoxylated glycerol of the instant invention is depicted by the following formula



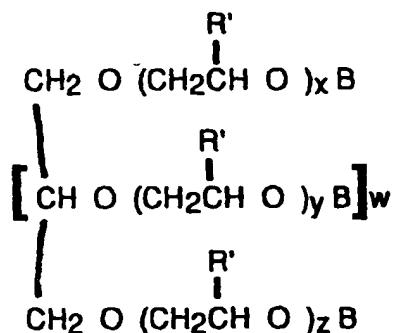
wherein w equals one to four, x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100.

[0065] In the dilute o/w microemulsion compositions or liquid crystal compositions the ethoxylated alcohol will be

present in admixture with the anionic surfactant. The proportion of the ethoxylated glycerol type based upon the weight of the liquid crystal composition or the final dilute o/w microemulsion composition will be 0.1% to 20% by weight.

[0066] Furthermore, in the more preferred compositions the weight ratio of nonsoap anionic surfactant to the ethoxylated polyhydric alcohol will be in the range of 3:1 to 1:3 with especially good results being obtained at a weight ratio of 2:1.

[0067] The instant composition can also contain 0 to 10 wt. %, of a monoester of an ethoxylated polyhydric alcohol depicted by the formula

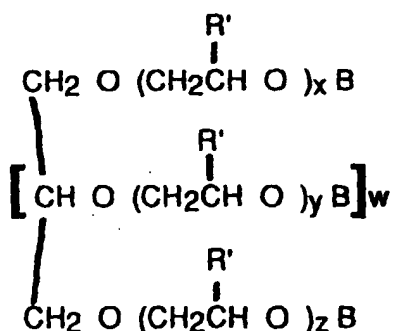


wherein w equals one to four. Two of the Bs are hydrogen and one B is selected from the group consisting of a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100.

[0068] The instant composition can also contain 0 to 2 wt. %, of a diester of an ethoxylated polyhydric alcohol depicted by the formula



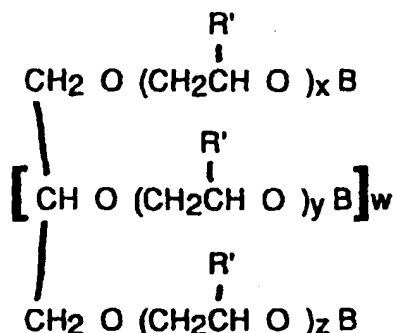
wherein w equals one to four. One of the Bs is hydrogen and two Bs are selected from the group consisting of a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having

6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals about 2 to 100.

[0069] The instant composition can also contain 0 to 1.0 wt. %, of a triester of an ethoxylated polyhydric alcohol depicted by the formula



wherein w equals one to four. The three Bs are selected from the group consisting of a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein a hydrogenated tallow alkyl chain or a coco alkyl chain is most preferred, x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100.

[0070] The instant compositions contain 0 to 1.0 wt. % of a tri-alkyl citrate such as tri-n-butyl citrate, tri-n-propyl citrate, tri-isopropyl citrate, tri-isobutyl citrate, tri-n-pentyl citrate, tri-isopentyl citrate and tri-n-hexyl, wherein tri-n-butyl citrate is preferred. The tri-n-butyl citrate functions in the formula as a foam control agent in that the foam is more readily collapsed such that the article can be rinsed more effectively.

[0071] The cosurfactant may play an essential role in the formation of the the liquid crystal composition or dilute o/w microemulsion and the concentrated microemulsion compositions. Three major classes of compounds have been found to provide highly suitable cosurfactants for the microemulsion over temperature ranges extending from 5°C to 43°C for instance; (1) water-soluble C<sub>3</sub>-C<sub>4</sub> alkanols, polypropylene glycol of the formula HO(CH<sub>2</sub>CHCH<sub>2</sub>O)<sub>n</sub>H wherein n is a number from 2 to 18 and monoalkyl ethers and esters of ethylene glycol and propylene glycol having the structural formulas R(X)<sub>n</sub>OH and R<sub>1</sub>(X)<sub>n</sub>OH wherein R is C<sub>1</sub>-C<sub>6</sub> alkyl, R<sub>1</sub> is C<sub>2</sub>-C<sub>4</sub> acyl group, X is (OCH<sub>2</sub>CH<sub>2</sub>) or (OCH<sub>2</sub>(CH<sub>3</sub>)CH) and n is a number from 1 to 4; (2) aliphatic mono- and di-carboxylic acids containing 2 to 10 carbon atoms, preferably 3 to 6 carbons in the molecule; and (3) triethyl phosphate. Additionally, mixtures of two or more of the four classes of cosurfactant compounds may be employed where specific pH's are desired.

[0072] When the mono- and di-carboxylic acid (Class 2) cosurfactants are employed in the instant microemulsion compositions at a concentration of 2 to 10 wt. %, the microemulsion compositions can be used as a cleaners for bathtubs and other hard surfaced items, which are acid resistant thereby removing lime scale, soap scum and greasy soil from the surfaces of such items damaging such surfaces. If these surfaces are of zirconium white enamel, they can be damaged by these compositions.

[0073] An aminoalkylene phosphonic acid at a concentration of 0.01 to 0.2 wt. % can be optionally used in conjunction with the mono- and di-carboxylic acids, wherein the aminoalkylene phosphonic acid helps prevent damage to zirconium white enamel surfaces. Additionally, 0.05 to 1% of phosphoric acid can be used in the composition.

[0074] Methanol and ethanol are explicitly excluded from the instant composition because of their low flash point.

[0075] Representative members of the polypropylene glycol include dipropylene glycol and polypropylene glycol having a molecular weight of 200 to 1000, e.g., polypropylene glycol 400. Other satisfactory glycol ethers are ethylene glycol monobutyl ether (butyl cellosolve), diethylene glycol monobutyl ether (butyl carbitol), triethylene glycol monobutyl ether, mono, di, tri propylene glycol monobutyl ether, tetraethylene glycol monobutyl ether, propylene glycol tertiary butyl ether, ethylene glycol monoacetate and dipropylene glycol propionate. When these glycol type cosurfactants are at a concentration of at least 1.0 weight %, in combination with a perfume at a concentration of at least 0.5 weight %, one can form a liquid crystal composition

[0076] Representative members of the aliphatic carboxylic acids include C<sub>3</sub>-C<sub>6</sub> alkyl and alkenyl monobasic acids and dibasic acids such as glutaric acid and mixtures of glutaric acid with adipic acid and succinic acid, as well as mixtures of the foregoing acids as well as acrylic acid or propionic acid.

[0077] While all of the aforementioned glycol ether compounds and acid compounds provide the described stability, the most preferred cosurfactant compounds of each type, on the basis of cost and cosmetic appearance (particularly odor), are diethylene glycol monobutyl ether and a mixture of adipic, glutaric and succinic acids, respectively. The ratio of acids in the foregoing mixture is not particularly critical and can be modified to provide the desired odor. Generally, to maximize water solubility of the acid mixture glutaric acid, the most water-soluble of these three saturated aliphatic dibasic acids, will be used as the major component.

[0078] Generally, weight ratios of adipic acid: glutaric acid:succinic acid is 1-3:1-8:1-5, can be used with equally good results.

[0079] Still other classes of cosurfactant compounds providing stable microemulsion compositions at low and elevated temperatures are the mono-, di- and triethyl esters of phosphoric acid such as triethyl phosphate.

[0080] The amount of cosurfactant required to stabilize the liquid crystal compositions or the microemulsion compositions will, of course, depend on such factors as the surface tension characteristics of the cosurfactant, the type and amounts of the primary surfactants and perfumes, and the type and amounts of any other additional ingredients which may be present in the composition and which have an influence on the thermodynamic factors enumerated above. The amounts of cosurfactant in the range of from 0.5% to 15%, by weight provide stable dilute o/w microemulsions for the above-described levels of primary surfactants and perfume and any other additional ingredients as described below.

[0081] As will be appreciated by the practitioner, the pH of the final microemulsion will be dependent upon the identity of the cosurfactant compound, with the choice of the cosurfactant being effected by cost and cosmetic properties, particularly odor. For example, microemulsion compositions which have a pH in the range of 1 to 10 may employ either the class 1 or the class 4 cosurfactant as the sole cosurfactant, but the pH range is reduced to 1 to 8.5 when the polyvalent metal salt is present. On the other hand, the class 2 cosurfactant can only be used as the sole cosurfactant where the product pH is below 3.2. However, where the acidic cosurfactants are employed in admixture with a glycol ether cosurfactant, compositions can be formulated at a substantially neutral pH (e.g., pH 7±1.5).

[0082] The ability to formulate neutral and acidic products without builders which have grease removal capacities is a feature of the present invention because the prior art o/w microemulsion formulations most usually are highly alkaline or highly built or both.

[0083] In addition to their excellent capacity for cleaning greasy and oily soils, the low pH o/w microemulsion formulations also exhibit excellent cleaning performance and removal of soap scum and lime scale in neat (undiluted) as well as in diluted usage.

[0084] The final essential ingredient in the inventive microemulsion compositions having improved interfacial tension properties is water. The proportion of water in the microemulsion compositions generally is in the range of 20% to 97%, by weight of the usual diluted o/w microemulsion composition.

[0085] As believed to have been made clear from the foregoing description, the dilute o/w microemulsion liquid all-purpose cleaning compositions of this invention are especially effective when used as is, that is, without further dilution in water, since the properties of the composition as an o/w microemulsion are best manifested in the neat (undiluted) form. However, at the same time it should be understood that depending on the levels of surfactants, cosurfactants, perfume and other ingredients, some degree of dilution without disrupting the microemulsion, per se, is possible. For example, at the preferred low levels of active surfactant compounds (i.e., primary anionic and nonionic surfactants) dilutions up to 50% will generally be well tolerated without causing phase separation, that is, the microemulsion state will be maintained.

[0086] However, even when diluted to a great extent, such as a 2- to 10-fold or more dilution, for example, the resulting compositions are still effective in cleaning greasy, oily and other types of soil. Furthermore, the presence of magnesium ions or other polyvalent ions, e.g., aluminum, as will be described in greater detail below further serves to boost cleaning performance of the primary detergents in dilute usage.

[0087] On the other hand, it is also within the scope of this invention to formulate highly concentrated microemulsions which will be diluted with additional water before use.

[0088] The present invention also relates to a stable concentrated microemulsion or acidic microemulsion composition comprising by weight:

- (a) 1 to 30% of an anionic surfactant;
- (b) 0.5 to 15% of an ethoxylated glycerol type compound;
- (c) 2 to 30% of a cosurfactant;
- (d) 0.4 to 10% of a water insoluble hydrocarbon or perfume;
- (e) 0 to 18% of at least one dicarboxylic acid;
- (f) 0 to 1% of phosphoric acid;

- (g) 0 to 0.2% of an aminoalkylene phosphonic acid;
- (h) 0 to 15% of magnesium sulfate heptahydrate;
- (i) 0 to 1.0% of a tri-alkyl citrate; and
- (h) balance being water, wherein the composition has an ecotoxicity as measured by the LC 50 test of at least 0.18 ml/L measured on Daphniae microorganisms.

[0089] The present invention also relates to a stable liquid crystal microemulsion or acidic microemulsion composition comprising by weight:

- (a) 1 to 30% of an anionic surfactant;
- (b) 0.5 to 15% of an ethoxylated glycerol type compound;
- (c) 0 to 2.5% of a fatty acid;
- (d) 2 to 30% of a cosurfactant;
- (e) 0.5 to 10% of a water insoluble hydrocarbon or perfume;
- (f) 0 to 15% of magnesium sulfate heptahydrate;
- (g) 0 to 1.0% of a tri-alkyl citrate; and
- (h) balance being water, wherein the composition has an ecotoxicity as measured by the LC 50 test of at least 0.18 ml/L measured on Daphniae microorganisms.

[0090] The present invention also relates to a stable concentrated microemulsion or acidic microemulsion composition comprising approximately by weight:

- (a) 1 to 30% of an anionic surfactant;
- (b) 0.5 to 15% of an ethoxylated polyhydric alcohol such as an ethoxylated glycerol;
- (c) 2 to 30% of a cosurfactant;
- (d) 0.4 to 10% of a water insoluble hydrocarbon or perfume;
- (e) 0 to 18% of at least one dicarboxylic acid;
- (f) 0 to 1% of phosphoric acid;
- (g) 0 to 0.2% of an aminoalkylene phosphonic acid;
- (h) 0 to 15% of magnesium sulfate heptahydrate; and
- (i) the balance being water.

[0091] The present invention also relates to a stable liquid crystal microemulsion or acidic microemulsion composition comprising approximately by weight:

- (a) 1 to 30% of an anionic surfactant;
- (b) 0.5 to 15% of an ethoxylated polyhydric alcohol such as an ethoxylated glycerol;
- (c) 2 to 30% of a cosurfactant;
- (d) 0.5 to 10% of a water insoluble hydrocarbon or perfume;
- (e) 0 to 15% of magnesium sulfate heptahydrate; and
- (f) the balance being water.

[0092] Such concentrated microemulsions can be diluted by mixing with up to 20 times or more, their weight of water to form o/w microemulsions similar to the diluted microemulsion compositions described above. While the degree of dilution is suitably chosen to yield an o/w microemulsion composition after dilution, it should be recognized that during the course of dilution both microemulsion and non-microemulsions may be successively encountered.

[0093] In addition to the above-described essential ingredients required for the formation of the liquid crystal composition or the microemulsion composition, the compositions of this invention may often and preferably do contain one or more additional ingredients which serve to improve overall product performance.

[0094] One such ingredient is an inorganic or organic salt of oxide of a multivalent metal cation, particularly  $Mg^{++}$ . The metal salt or oxide provides several benefits including improved cleaning performance in dilute usage, particularly in soft water areas, and minimized amounts of perfume required to obtain the microemulsion state. Magnesium sulfate, either anhydrous or hydrated (e.g., heptahydrate), is especially preferred as the magnesium salt. Good results also have been obtained with magnesium oxide, magnesium chloride, magnesium acetate, magnesium propionate and magnesium hydroxide. These magnesium salts can be used with formulations at neutral or acidic pH since magnesium hydroxide will not precipitate at these pH levels.

[0095] Although magnesium is the preferred multivalent metal from which the salts (inclusive of the oxide and hydroxide) are formed, other polyvalent metal ions also can be used provided that their salts are nontoxic and are sol-

uble in the aqueous phase of the system at the desired pH level. Thus, depending on such factors as the pH of the system, the nature of the primary surfactants and cosurfactant, and so on, as well as the availability and cost factors, other suitable polyvalent metal ions include aluminum, copper, nickel, iron, calcium, etc. It should be noted, for example, that with the preferred paraffin sulfonate anionic detergent calcium salts will precipitate and should not be used. It has also been found that the aluminum salts work best at pH below 5 or when a low level, for example 1 weight percent, of citric acid is added to the composition which is designed to have a neutral pH. Alternatively, the aluminum salt can be directly added as the citrate in such case. As the salt, the same general classes of anions as mentioned for the magnesium salts can be used, such as halide (e.g., bromide, chloride), sulfate, nitrate, hydroxide, oxide, acetate, propionate, etc.

[0096] In the dilute compositions the metal compound is added to the composition in an amount sufficient to provide at least a stoichiometric equivalence between the anionic surfactant and the multivalent metal cation. For example, for each gram-ion of  $Mg^{++}$  there will be 2 gram moles of paraffin sulfonate, alkylbenzene sulfonate, etc., while for each gram-ion of  $Al^{3+}$  there will be 3 gram moles of anionic surfactant. Thus, the proportion of the multivalent salt generally will be selected so that one equivalent of compound will neutralize from 0.1 to 1.5 equivalents, of the acid form of the anionic surfactant.

[0097] At higher concentrations of anionic surfactant, the amount of multivalent salt will be in range of 0.5 to 1 equivalents per equivalent of anionic surfactant.

[0098] The liquid crystal composition or the o/w microemulsion compositions comprises from 0% to 2.5%, by weight of the composition of a C8-C<sub>22</sub> fatty acid or fatty acid soap as a foam suppressant. The addition of fatty acid or fatty acid soap provides an improvement in the rinseability of the composition whether applied in neat or diluted form. Generally, however, it is necessary to increase the level of cosurfactant to maintain product stability when the fatty acid or soap is present. If more than 2.5wt % of the fatty acid is used in the instant compositions, the composition will become unstable at low temperatures as well as having an objectionable smell.

[0099] As example of the fatty acids which can be used as such or in the form of soap, mention can be made of distilled coconut oil fatty acids, "mixed vegetable" type fatty acids (e.g. high percent of saturated, mono-and/or polyunsaturated C<sub>18</sub> chains); oleic acid, stearic acid, palmitic acid, eicosanoic acid, and the like, generally those fatty acids having from 8 to 22 carbon atoms being acceptable.

[0100] The all-purpose liquid cleaning composition of this invention may, if desired, also contain other components either to provide additional effect or to make the product more attractive to the consumer. The following are mentioned by way of example: Colors or dyes in amounts up to 0.5% by weight; bactericides in amounts up to 1% by weight; preservatives or antioxidizing agents, such as formalin, 5-bromo-5-nitro-dioxan-1,3; 5-chloro-2-methyl-4-isothiazolin-3-one, 2,6-di-tert.butyl-p-cresol, etc., in amounts up to 2% by weight; and pH adjusting agents, such as sulfuric acid or sodium hydroxide, as needed. Furthermore, if opaque compositions are desired, up to 4% by weight of an opacifier may be added.

[0101] The instant compositions of the instant invention explicitly exclude zwitterionic surfactant such as betaines because these zwitterionic surfactants are extremely high foaming which, if used in the instant composition, would cause the instant compositions to have too high a foam profile and that too much foam would leave residue on the surface being cleaned.

[0102] In final form, the all-purpose liquids are clear oil-in-water microemulsions or liquid crystal compositions and exhibit stability at reduced and increased temperatures. More specifically, such compositions remain clear and stable in the range of 5°C to 50°C. Such compositions exhibit a pH in the acid or neutral range depending on intended end use. The liquid microemulsion compositions are readily pourable and exhibit a viscosity in the range of 6 to 60 millipascal . second (mPas.) as measured at 25°C. with a Brookfield RVT Viscometer using a #1 spindle rotating at 20 RPM.

[0103] The compositions are directly ready for use or can be diluted as desired and in either case no or only minimal rinsing is required and substantially no residue or streaks are left behind. Furthermore, because the compositions are free of detergent builders such as alkali metal polyphosphates they are environmentally acceptable and provide a better "shine" on cleaned hard surfaces.

[0104] When intended for use in the neat form, the liquid compositions can be packaged under pressure in an aerosol container or in a pump-type sprayer for the so-called spray-and-wipe type of application.

[0105] Because the compositions as prepared are aqueous liquid formulations and since no particular mixing is required to form the o/w microemulsion, the compositions are easily prepared simply by combining all the ingredients in a suitable vessel or container. The order of mixing the ingredients is not particularly important and generally the various ingredients can be added sequentially or all at once or in the form of aqueous solutions of each or all of the primary detergents and cosurfactants can be separately prepared and combined with each other and with the perfume. The magnesium salt, or other multivalent metal compound, when present, can be added as an aqueous solution thereof or can be added directly. It is not necessary to use elevated temperatures in the formation step and room temperature is sufficient.

[0106] The instant microemulsion formulas explicitly exclude alkali metal silicates and alkali metal builders such as alkali metal polyphosphates, alkali metal carbonates, alkali metal phosphonates and alkali metal citrates because these

materials, if used in the instant composition, would cause the composition to have a high pH as well as leaving residue on the surface being cleaned.

[0107] It is contemplated within the scope of the instant invention that the ethoxylated glycerol type compound can be employed in hard surface cleaning compositions such as wood cleaners, window cleaners and light duty liquid cleaners, wherein improvements in a grease release effect in desirable.

[0108] The following examples illustrate liquid cleaning compositions of the described invention. Unless otherwise specified, all percentages are by weight. The exemplified compositions are illustrative only and do not limit the scope of the invention. Unless otherwise specified, the proportions in the examples and elsewhere in the specification are by weight.

#### Example 1

[0109] The following compositions in wt. % were prepared:

	A	B	C	D	E	F	Mr. Proper	St Marc Lemon
Sodium C13-C17 Paraffin sulfonate	4.7	4.3	4	4.3	14.1	7.05	2.9	-
EO/PO nonionic	-	-	-	-	-	-	-	3.2
Levenol F-200	2.3	2.2	2	2.2	6.3	3.45	-	-
C13-C15 EO 14 nonionic	-	-	-	-	-	-	3.3	-
DEGMBE	4	4	3.5	4	12	6	4.4	3
Fatty acid	0.75	0.5	0.4	0.75	2.25	1.125	0.65	0.3
MgSO4 7 H2O	2.2	2	1.9	2.2	6.3	3.15	-	-
Perfume (a)	0.8	0.75	0.9	0.7	2.4	1.2	present	present
Sodium Citrate	-	-	-	-	-	-	3.2	-
Water	Bal	Bal	Bal	Bal	Bal	Bal	Bal	Bal
pH	7	7	7	7	7	7	9.5	7
Degreasing test								
Neat (b)	30	35	35	35	30	30	70	>100
Dilute (b)	45	60	60	60	45	45	>90	90
Residue	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.	Worse	Equal to ref.
Foam in hard Water	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.	Equal to ref.
LC50-Ecotoxicity on Daphniae (c)	0.18 ml/l	-	-	-	-	-	0.1 ml/l	0.033 ml/l

	G	H	I
Linear alkyl benzene sulfonic acid	4.7	4.5	5
NaOH to reach pH 7	0.6	0.57	0.64
Levenol F-200	2.3	2.5	2
DEGMBE	5.6	6	6.2
Fatty acid	0.75	0.75	0.75
MgSO4 7 H2O	2.15	2.06	2.3
Perfume (a)	0.8	0.8	0.8
Water	Bal	Bal	Bal
pH	7	7	7

(a) contains 25% by weight of terpenes.

(b) the lower the number of strokes, the better the degreasing performance.

(c) the higher the results, the lower the ecotoxicity.

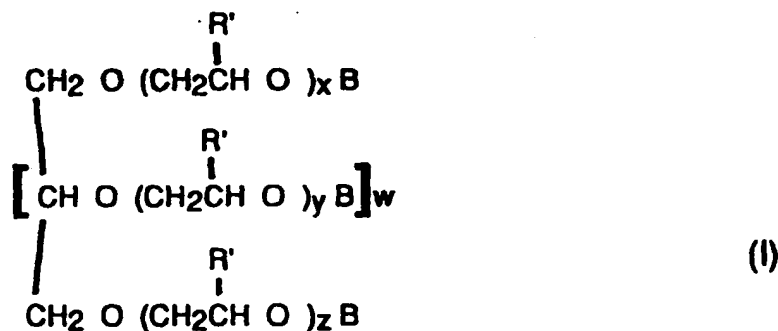
[0110] Furthermore, "dissolution power" of the o/w microemulsion of this example is compared to the "dissolution power" of an identical composition except that an equal amount (5 weight percent) of sodium cumene sulfonate hydrotrope is used in place of the diethylene glycol monobutyl ether cosurfactant in a test wherein equal concentrations of heptane are added to both compositions. The o/w microemulsion of this invention solubilizes 12 grams of the water immiscible substance as compared to 1.4 grams in the hydrotrope containing liquid composition.

[0111] In a further comparative test using blue colored cooking oil—a fatty triglyceride oil —, the composition of Example 1 is clear after the addition of 0.2 grams of cooking oil whereas the cooking oil floats on the top of the composition containing the sulfonate hydrotrope.

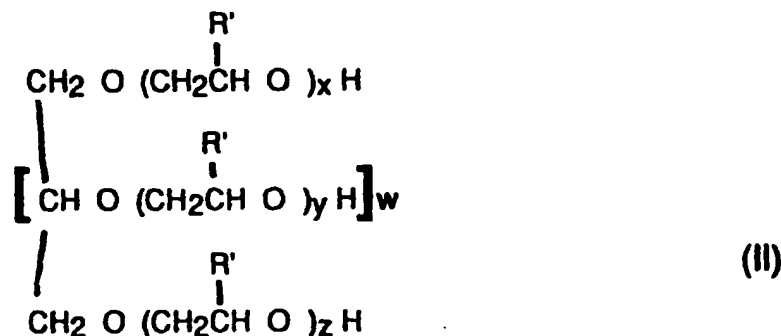
[0112] When the concentration of perfume is reduced to 0.4% in the composition of Example 1, a stable o/w microemulsion composition is obtained. Similarly, a stable o/w microemulsion is obtained when the concentration of perfume is increased to 2% by weight and the concentration of cosurfactant is increased to 6% by weight in Example 1.

[0113] The present invention also relates to an aqueous solution of a grease release system which comprises:

(a) 0.1 to 20.0 wt. % of a mixture of:



and



wherein w equals one to four. B is selected from the group consisting of hydrogen or a group represented by:



wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein at least one of the B groups is represented by said



R' is selected from the group consisting of hydrogen and methyl groups; x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100, wherein in Formula (I) the ratio of monoester / diester / triester is 40 to 90 / 5 to 35 / 1 to 20, wherein the ratio of Formula (I) to Formula (II) is a value between 3 to 0.02, more preferably 3 to 0.1, and most preferably 1.5 to 0.2, wherein it is most preferred that there is more of Formula (II) than Formula (I) in the mixture; and

(b) the balance being water.

[0114] The aqueous solution of the grease release agent can be coated onto a hard surface. The treated hard surface will prevent the adhering of grease to the hard surface thereby providing easier cleaning of the hard surface with a conventional hard surface cleaning composition.

#### Example 2

[0115] The example illustrates a typical formulation of a "concentrated" o/w microemulsion based on the present invention:

	% by weight
Coco fatty acid	4
Sodium C <sub>13</sub> -C <sub>17</sub> Paraffin Sulfonate	20.75
Levenol F-200	12
Diethylene glycol monobutyl ether	20
Perfume (a)	12.5
Water	Bal to 100
pH: 7.0 ± 0.2	

[0116] This concentrated formulation can be easily diluted, for example, five times with tap water, to yield a diluted o/w microemulsion composition. Thus, by using microemulsion technology it becomes possible to provide a product having high levels of active detergent ingredients and perfume, which has high consumer appeal in terms of clarity, odor and stability, and which is easily diluted at the usual usage concentration for similar all-purpose hard surface liquid cleaning compositions, while retaining its cosmetically attractive attributes.

[0117] Naturally, these formulations can be used, where desired, without further dilution and can also be used at full or diluted strength to clean soiled fabrics by hand or in an automatic laundry washing machine.

#### Example 3

[0118] This example illustrates a diluted o/w microemulsion composition according to the invention having an acidic pH and which also provides improved cleaning performance on soap scum and lime scale removal as well as for cleaning greasy soil.

	% by weight
Sodium C <sub>13</sub> -C <sub>17</sub> paraffin sulfonate	4.7
Levenol F-200	2.3
Mg SO <sub>4</sub> 7H <sub>2</sub> O	2.2
Mixture of succinic acid/glutaric acid/adipic acid (1:1:1)	5
Perfume (d)	1.0
Water, minors (dye)	balance to 100
Phosphonic acid	0.2
Amino tris - (methylene-phosphonic acid)	0.03
pH = 3 ± 0.2	

(d) contains 40% by weight of terpene

#### Example 4

[0119] Formula A of Example I was tested for the removal of a combination of grease and particulate soil as well as for a grease release effect and compared to commercial Ajax<sup>™</sup>NME

I. Grease + particulate soil removal;

#### Test Method

#### A) Soil composition:

[0120]

70 g of mineral oil  
35 g of particulate soil (vacuum cleaner dust + 1% of carbon black)  
35 g C<sub>2</sub>Cl<sub>4</sub>

#### B) Soil preparation:

[0121]

- Weigh cleaned/dried glass tiles
- Soil the tiles with the grease + particulate soil
- Bake the tiles 1 hour at 80°C
- Weigh the soiled tiles which aged 2 hours at RT.

#### C) Soil removal:

[0122]

The soiled tiles are soaked for 15 minutes at RT in the test products, then they are delicately rinsed with tap water. After drying 45 minutes at 50°C, the tiles are weighed again.

Results

[0123]

5

10

		Grease + particulate soil
% of removal mean of 6 tiles		
Commercial Ajax <sup>tm</sup> NME		60
Formula A of Example I		95

[0124] Formula A exhibits improved grease + particulate soil removal over the Commercial Ajax<sup>tm</sup> NME

15

## II. Grease release effect

Test Method

20

A) Soil composition:

[0125]

25

20% hardened tallow  
80% beef tallow  
fat blue dye

B) Soil preparation:

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[0126]

The fat mixture is heated and sprayed with an automatic spraying device on cleaned and dried ceramic tiles.

C) Soil removal:

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[0127]

Product used neat: 2.5 g on sponge

Product used dilute: 1.2% sol in tap water - 10 ml of the solution on the sponge

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The cleaning procedure is done with the gardner device for both product concentrations.

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ResultsA) On treated ceramic tiles (treated with the product before spraying the soil)

5 [0128]

	Neat	Dilute
	Number of Strokes	
First grease layer deposition	mean of 4 tiles	mean of 6 tiles
Commercial Current Ajax <sup>tm</sup> NME	27	19
	19	5*
Second grease layer deposition on the same tile	mean of 4 tiles	mean of 6 tiles
Commercial Ajax <sup>tm</sup> NME	25	48
	25	18*

\* highly significant difference

B) On untreated ceramic tiles

25

[0129] In addition to the previous test, the 3 following procedures were used to verify that Formula A remains on the surface after rinsing or wiping. After the first cleaning procedure and before the second spraying:

- 1) the tiles were allowed to dry in open air
- 2) the surface was wiped with paper towel
- 3) the surface was rinsed with wet sponge

30

- 1) dry in open air

35 [0130]

	Neat	Dilute
	Number of Strokes	
First grease layer deposition	mean of 4 tiles	mean of 6 tiles
Commercial Ajax <sup>tm</sup> NME	29	30
Formula A	27	32
Second grease layer deposition on the same tile	mean of 4 tiles	mean of 6 tiles
Commercial Ajax <sup>tm</sup> NME	33	21
Formula A	30	6*

\* highly significant difference

55

2) wipe dry the surface

[0131]

5

10

15

	Neat	Dilute
	Number of Strokes	
First grease layer deposition	mean of 4 tiles	mean of 6 tiles
Commercial Ajax <sup>tm</sup> NME	29	30
Formula A	27	32
Second grease layer deposition on the same tile	mean of 4 tiles	mean of 6 tiles
Commercial Ajax <sup>tm</sup> NME	35	46
Formula A	30	48.5

20 3) wet wiping the surface

[0132]

25

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35

	Neat	Dilute
	Number of Strokes	
First grease layer deposition	mean of 4 tiles	mean of 6 tiles
Commercial Ajax <sup>tm</sup> NME	29	30
Formula A	27	32
Second grease layer deposition on the same tile	mean of 4 tiles	mean of 6 tiles
Commercial Ajax <sup>tm</sup> NME	34	58
Formula A	27	41 **

\*\* after 5 strokes, 65% of the grease is already removed

40

[0133] These results clearly demonstrate the important grease release effect obtained with Formula A especially when the product is used dilute.

#### Example 5

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[0134] The following liquid crystal compositions were prepared by simple mixing procedure

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55

	A	B	C
Sodium C <sub>13</sub> -C <sub>17</sub> Paraffin sulfonate	4.3	4.3	4.3
Levenol F-200	2.2	2.2	2.2
Propylene glycol monbutyl ether	3.5	-	-
Dipropylene glycol monbutyl ether	-	3.5	-
Tripropylene glycol monbutyl ether	-	-	3.5

(continued)

	A	B	C
Fatty acid	0.5	0.5	0.5
MgSO <sub>4</sub> 7 H <sub>2</sub> O	1.6	1.6	1.6
Perfume (a)	1	1.5	1.5
Water	Bal	Bal	Bal
pH	7	7	7

**Example 6**

[0135] The following optically clear microemulsion compositions were made by forming first a solution by mixing at 25°C water, magnesium lauryl ether sulfate, Levenol V-510/2 and 1-Pentanol. To this solution with mixing at 25°C was added the dodecane to form the optically clear microemulsion. The formula are expressed in weight percent.

	A	B	C	D	E	F	G	H	I
Magnesium Lauryl sulfate	7	2.04	3.04	4.99	3.01	6.38	5.01	4.02	2.99
Levenol V-501/2	3.2	8.15	7.1	5.1	7.06	3.9	5.06	6.24	7.2
1-Pentanol	1.19	1.03	4.1	4.05	5.05	5.67	1.07	1.05	1.13
Dodecane	1.29	0.73	17.36	11.26	20.07	15.2	2.86	3.05	2.9
water	Bal	Bal	Bal	Bal	Bal	Bal	Bal	Bal	Bal

**Example 7**

[0136] The following composition was prepared by simple mixing procedure:

Sodium C <sub>13</sub> -C <sub>17</sub>	4.0
Paraffin Sulfonate	
Levenol F-200	2.0
DEGMBE	4.5
Fatty Acid	0.5
MgSO <sub>4</sub> ·7H <sub>2</sub> O	1.8
Perfume (a)	0.8
tri-n-butyl citrate	0.5
Water	Balance
pH	7
Degreasing test	
Neat (6)	
Dilute (6)	
Residue	
Foam in hand	
Water	

(a) contains 25% by weight of terpenes

(b) the lower the number of strokes, the better the degreasing performance.

[0137] The addition of the tri-n-butyl citrate improves the rinsability of the surface being rinsed in that the collapse of the foam is improved as compared to a composition not containing the tri-n-butyl citrate.

[0138] In summary, the described invention broadly relates to an improvement in microemulsion compositions containing an anionic surfactant, an ethoxylated glycerol type compound, a fatty acid, one of the specified cosurfactants, a hydrocarbon ingredient and water which comprise the use of a water-insoluble, odoriferous perfume as the essential hydrocarbon ingredient in a proportion sufficient to form either a dilute o/w microemulsion composition or liquid crystal composition containing, by weight, 0.1% to 20.0% of an anionic detergent, 0.1% to 10% of an ethoxylated glycerol type compound, 0% to 50% of cosurfactant, 0 to 1.0% of a tri-alkyl citrate, 0.4% to 10% of perfume and the balance being water.

#### Example 8

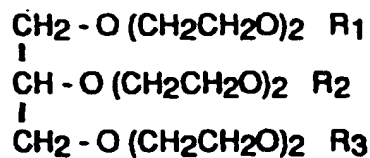
[0139] The following compositions in wt. % were prepared:

	A	B	C	D	E	F
Na C13-17 paraffin sulfonate	4.7	4.7	4.7	4.7	4.7	4.7
DEGMBE	4	4	4	4	4	4
Coco fatty acid	0.75	0.75	0.75	0.75	0.75	0.75
MgSO <sub>4</sub>	2.2	2.2	2.2	2.2	2.2	2.2

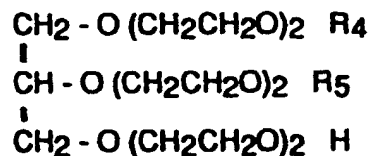
(continued)

	A	B	C	D	E	F
Perfume	0.8	0.8	0.8	0.8	0.8	0.8
Compound (a)	0.023	0.017	0.011	0.006	--	--
Compound (b)	0.115	0.086	0.058	0.028	--	--
Compound (c)	0.897	0.673	0.449	0.224	--	--
Compound (d)	1.265	1.525	1.78	2.066	2.3	--
Neodol 91-5	--	--	--	--	--	2.3
Water	Bal.	Bal.	Bal.	Bal.	Bal.	Bal.
Phase behavior	One phase	One phase	One phase	One phase	One phase	One phase
Particulate soil (Kaolin removal)	71.0	79.8	84.0	86.0	88.7	51.0

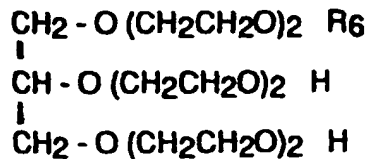
Compound (a) is



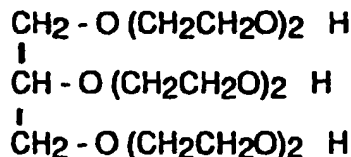
wherein  $\text{R}_1$ ,  $\text{R}_2$  and  $\text{R}_3$  are coco alkyl chains  
Compound (b) is



wherein  $\text{R}_4$  and  $\text{R}_5$  are coco alkyl chains  
Compound (c) is



wherein  $\text{R}_6$  is a coco alkyl chains  
Compound (d) is

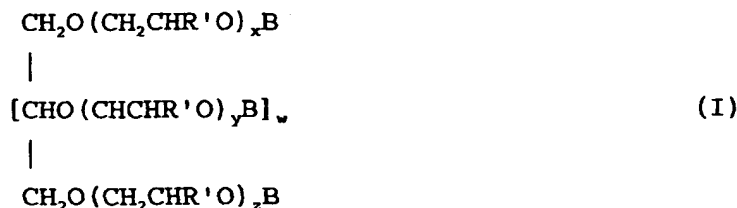


[0140] In summary, the described invention broadly relates to an improvement in microemulsion compositions for the removal of particulate soil containing an anionic surfactant, an ethoxylated polyhydric alcohol a cosurfactant, a hydrocarbon ingredient and water which comprise the use of a water-insoluble, odoriferous perfume as the essential hydrocarbon ingredient in a proportion sufficient to form either a dilute o/w microemulsion composition or liquid crystal composition containing, by weight, 0.1% to 20% of an anionic detergent, 0.1% to 20.0% of an ethoxylated polyhydric alcohol, 0% to 50% of cosurfactant, 0.4% to 10% of perfume and the balance being water.

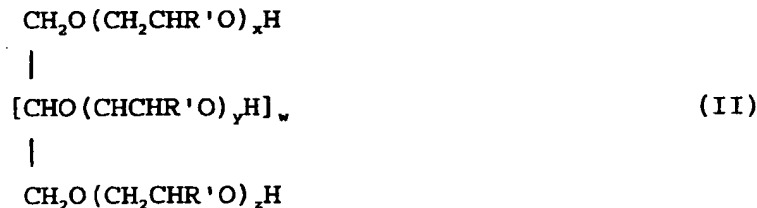
[0141] In summary, the described invention broadly relates to an improvement in microemulsion compositions containing an anionic surfactant, an esterified polyethoxyether nonionic surfactant, a fatty acid, one of the specified cosurfactants, a hydrocarbon ingredient and water which comprise the use of a water-insoluble, odoriferous perfume as the essential hydrocarbon ingredient in a proportion sufficient to form either a dilute o/w microemulsion composition or liquid crystal composition containing, by weight, 0.1% to 20% of an anionic detergent, 0.1% to 20.0% of a solubilizing agent which is an ethoxylated glycerol type compound, 0% to 50% of cosurfactant, 0.4% to 10% of perfume and the balance being water.

#### Claims

1. Liquid crystal composition having a viscosity at 25°C of 6 to 60 mPa · s and comprising water and by weight: 0.1% to 20% of an anionic surfactant; 1% to 50% of a cosurfactant; 0% to 2.5% of a fatty acid; 0.1% to 20% of a mixture of



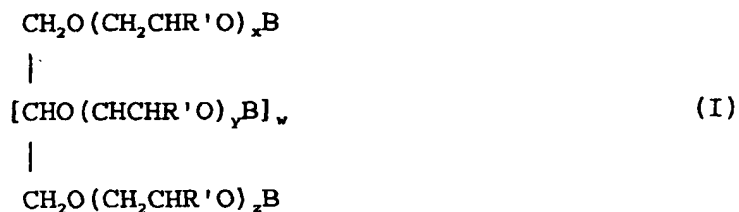
and



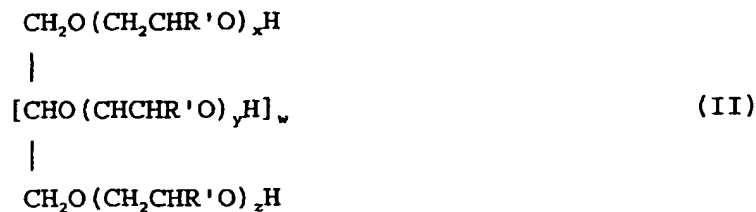
wherein w equals 1-4, B is selected from the group consisting of hydrogen or a group represented by CO-R, wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein at least one of the B groups is represented by said CO-R, R' is selected from the group consisting of hydrogen and methyl groups; x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100, wherein in Formula (I) the ratio of monoester / diester / triester is 45 to 90 / 5 to 40 / 1 to 20, wherein

the ratio of Formula (I) to Formula (II) is a value between 3 to 0.02; 0.1% to 10% of a water insoluble hydrocarbon or a perfume; and the balance being water, wherein said composition does not contain a zwitterionic surfactant or methanol.

- 5 2. Liquid crystal composition according to claim 1, which further comprises 0.4% to 1.0% of a tri-alkyl citrate.
3. Liquid crystal composition according to any one of the previous claims, which further comprises a salt of a multivalent metal cation in an amount sufficient to provide from 0.5 to 1.5 equivalents of said cation per equivalent of said anionic surfactant.
- 10 4. Liquid crystal composition according to claim 3 wherein the multivalent metal cation is magnesium or aluminium.
5. Liquid crystal composition according to claim 3 or 4, wherein said composition contains 0.9 to 1.4 equivalents of said cation per equivalent of anionic detergent.
- 15 6. Liquid crystal composition according to claim 3 to 5, wherein said multivalent salt is magnesium oxide, magnesium chloride or magnesium sulfate.
7. Liquid crystal composition according to any of the previous claims, further comprising fatty acid having 8 to 22 carbon atoms.
- 20 8. Liquid crystal composition according to any of the previous claims, which contains from 0.5% to 15% by weight of said cosurfactant and from 0.4% to 3.0% by weight of said hydrocarbon.
- 25 9. Liquid crystal composition according to any of the previous claims, wherein the cosurfactant is a water soluble glycol ether.
10. Liquid crystal composition according to claim 9, wherein the glycol ether is selected from the group consisting of ethylene glycol monobutyl ether, diethylene glycol monobutyl ether, triethylene glycol monobutyl ether, poly-propylene glycol having an average molecular weight of from 200 to 1,000 and propylene glycol tert.butyl ether, mono, di, tri propylene glycol monobutyl ether.
- 30 11. Liquid crystal composition according to claim 10, wherein the glycol ether is ethylene glycol monobutyl ether or diethylene glycol monobutyl ether.
- 35 12. Liquid crystal composition according to any of the previous claims, wherein the cosurfactant is a C<sub>3</sub>-C<sub>6</sub> aliphatic carboxylic acid selected from the group consisting of acrylic acid, propionic acid, glutaric acid, mixtures of glutaric acid and succinic acid and adipic acid and mixtures of any of the foregoing.
- 40 13. Liquid crystal composition according to claim 12 wherein the aliphatic carboxylic acid is a mixture of adipic acid, glutaric acid and succinic acid.
14. Liquid crystal composition according to any of the previous claims, wherein the anionic surfactant is a C<sub>9</sub>-C<sub>15</sub> alkyl benzene sulfonate or a C<sub>10</sub>-C<sub>20</sub> alkane sulfonate.
- 45 15. Grease release system which comprises:
  - (a) 0.1% to 20% of a mixture of



and



wherein w equals 1-4, B is selected from the group consisting of hydrogen or a group represented by CO-R, wherein R is selected from the group consisting of alkyl group having 6 to 22 carbon atoms, and alkenyl groups having 6 to 22 carbon atoms, wherein at least one of the B groups is represented by said CO-R, R' is selected from the group consisting of hydrogen and methyl groups; x, y and z have a value between 0 and 60, provided that (x+y+z) equals 2 to 100, wherein in Formula (I) the ratio of monoester / diester / triester is 40 to 90 / 5 to 35 / 1 to 20, wherein the ratio of Formula (I) to Formula (II) is a value between 3 to 0.02; (b) the balance being water.

16. Grease release system according to claim 15, wherein the ratio of Formula (I) to Formula (II) is a value between 3 to 0.1, preferably between 1.5 to 0.2.

17. Grease release system according to claim 15 or 16, wherein there is more of Formula (II) than of Formula (I) in the mixture.



European Patent  
Office

## EUROPEAN SEARCH REPORT

Application Number  
EP 99 20 4486

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
P,X	EP 0 637 629 A (COLGATE-PALMOLIVE CO.) 8 February 1995 (1995-02-08) * the whole document *	1,3-17	C11D17/00 C11D1/825 C11D1/83 C11D3/20
P,X	WO 95 14765 A (COLGATE-PALMOLIVE CO.) 1 June 1995 (1995-06-01) * page 13, line 1 - page 15, line 16 * * claims *	1,3-17	//C11D1:72, C11D1:74, C11D1:22, C11D1:14
P,X	EP 0 672 747 A (COLGATE-PALMOLIVE CO.) 20 September 1995 (1995-09-20) * page 6, line 45 - page 10, line 27 * * page 12, line 23 - line 30 * * claims *	1,3-17	
P,X	EP 0 668 346 A (COLGATE-PALMOLIVE CO.) 23 August 1995 (1995-08-23) * page 5, line 16 - page 6, line 46 * * claims *	1,3-11, 15-17	
P,X	EP 0 677 578 A (COLGATE-PALMOLIVE CO.) 18 October 1995 (1995-10-18) * page 5, line 23 - page 9, line 18 * * page 11, line 3 - line 8 * * claims 1-6 *	1,3-11, 15-17	TECHNICAL FIELDS SEARCHED (Int.Cl.7)  C11D
D,X	GB 1 453 385 A (KAO SOAP) 20 October 1976 (1976-10-20) * claims *	15	
A		1	
D,A	GB 2 190 681 A (COLGATE-PALMOLIVE CO.) 25 November 1987 (1987-11-25) * claims *	1,2,14	
D,A	GB 2 144 763 A (THE PROCTER & GAMBLE CO.) 13 March 1985 (1985-03-13) * page 2, line 16 - line 44 * * claims *	1,3-11, 14	
The present search report has been drawn up for all claims			
Place of search <b>THE HAGUE</b>		Date of completion of the search <b>2 February 2000</b>	Examiner <b>Serbetsoglou, A</b>
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			

EPO FORM 1503 03/82 (P04C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

EP 99 20 4486

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on  
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02-02-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0637629 A	08-02-1995	AU 691499 B	21-05-1998
		AU 6878094 A	16-02-1995
		BR 9403153 A	11-04-1995
		CA 2129399 A	05-02-1995
		HU 67925 A	20-03-1995
		NO 942886 A	06-02-1995
		NZ 264113 A	25-06-1996
		PL 304558 A	06-02-1995
		PT 101556 A	04-05-1995
		RO 114905 A	30-08-1999
		US 5952281 A	14-09-1999
		US 5985814 A	16-11-1999
		US 5599785 A	04-02-1997
		US 5549840 A	27-08-1996
		US 5716925 A	10-02-1998
		US 5610130 A	11-03-1997
		US 5763386 A	09-06-1998
		US 5759983 A	02-06-1998
		US 5741760 A	21-04-1998
		US 5776880 A	07-07-1998
		US 5731281 A	24-03-1998
		US 5861367 A	19-01-1999
		US 5854193 A	29-12-1998
		US 5942482 A	24-08-1999
		ZA 9405565 A	29-01-1996
WO 9514765 A	01-06-1995	AU 680087 B	17-07-1997
		AU 1289195 A	13-06-1995
		BR 9408122 A	05-08-1997
		EP 0730636 A	11-09-1996
		NZ 277602 A	22-09-1997
		PL 314622 A	16-09-1996
		US 5716925 A	10-02-1998
		US 5854193 A	29-12-1998
EP 0672747 A	20-09-1995	AU 1355995 A	21-09-1995
		BR 9501075 A	04-06-1996
		CA 2144367 A	15-09-1995
		HU 70070 A	28-09-1995
		NZ 270633 A	24-06-1997
		US 5763386 A	09-06-1998
EP 0668346 A	23-08-1995	US 5571459 A	05-11-1996
		AU 680076 B	17-07-1997
		AU 1148995 A	17-08-1995
		BR 9500451 A	26-09-1995

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

EP 99 20 4486

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

02-02-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0668346 A		CA 2141926 A	08-08-1995
		HU 70071 A	28-09-1995
		NZ 270433 A	25-06-1996
		US 5561106 A	01-10-1996
		PL 307113 A	10-06-1996
EP 0677578 A	18-10-1995	AU 681487 B	28-08-1997
		AU 1502095 A	26-10-1995
		BR 9501578 A	14-11-1995
		CA 2146068 A	16-10-1995
		JP 8081693 A	26-03-1996
		US 5610130 A	11-03-1997
		ZA 9502387 A	23-09-1996
GB 1453385 A	20-10-1976	JP 1003295 C	27-06-1980
		JP 50089405 A	17-07-1975
		JP 54038603 B	22-11-1979
		DE 2457927 A	12-06-1975
GB 2190681 A	25-11-1987	US 5075026 A	24-12-1991
		AT 957 U	26-08-1996
		AU 597367 B	31-05-1990
		AU 7313887 A	26-11-1987
		BE 1001742 A	27-02-1990
		BR 8702610 A	23-02-1988
		CA 1302194 A	02-06-1992
		CH 676854 A	15-03-1991
		DE 3716526 A	26-11-1987
		DK 258987 A	22-11-1987
		ES 2004934 A	16-02-1989
		FR 2599046 A	27-11-1987
		GR 870802 A	09-09-1987
		HK 60094 A	08-07-1994
		LU 86888 A	22-11-1988
		MX 169901 B	30-07-1993
		NL 8701215 A	16-12-1987
		NO 872116 A, B,	23-11-1987
		NZ 220271 A	29-01-1990
		PH 26507 A	07-08-1992
		PT 84886 A, B	01-06-1987
		SE 503826 C	16-09-1996
		SE 8702084 A	22-11-1987
		SG 20994 G	10-06-1994
		TR 25009 A	28-08-1992
		US 5082584 A	21-01-1992
		US 5076954 A	31-12-1991

EPO FORM P0458

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.

EP 99 20 4486

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

02-02-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
GB 2144763 A	13-03-1985	NONE	

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82